

SREE CHAITANYA MAHAVIDYALAYA

(Affiliated by WEST BENGAL STATE UNIVERSITY)

Internal Examination, 2021 – Sem. IV

B.Sc. - General

PAPER-CODE: CEMGCOR04T

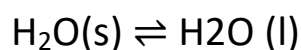
PAPER-NAME: Chemistry

Time: 30 Minutes

Full Marks: 8

A. Answer the following questions: (any four) (2 X 4 = 8)

1. What is Azeotropic mixture? Give one example.
2. Calculate the degrees of freedom (F) of the following equation.



3. At 25°C temperature the equivalent conductance at infinite dilution for AgNO_3 , KNO_3 and KCl solutions are 133.36, 144.96 and 149.86 $\text{ohm}^{-1}\text{cm}^2\text{gmeq}^{-1}$ respectively. Find out the equivalent conductance at infinite dilution for AgCl .
4. Construct the cell of the following cell reaction
$$\text{Cu(s)} + 2\text{Ag}^+(\text{aq}) \longrightarrow \text{Cu}^{2+}(\text{aq}) + 2\text{Ag(s)}$$
5. What is primary standard substance? Give one example.

CLICK THE LINK BELOW TO SUBMIT ANSWER-SCRIPT (In single PDF)

<https://forms.gle/5TVZu6n9M8XCT3936>